

# Enhanced metal-ion adsorption and diffusion in two-dimensional C<sub>3</sub>N/Black phosphorene heterostructures: A multidisciplinary computational-experimental approach for next-generation batteries

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## Introduction

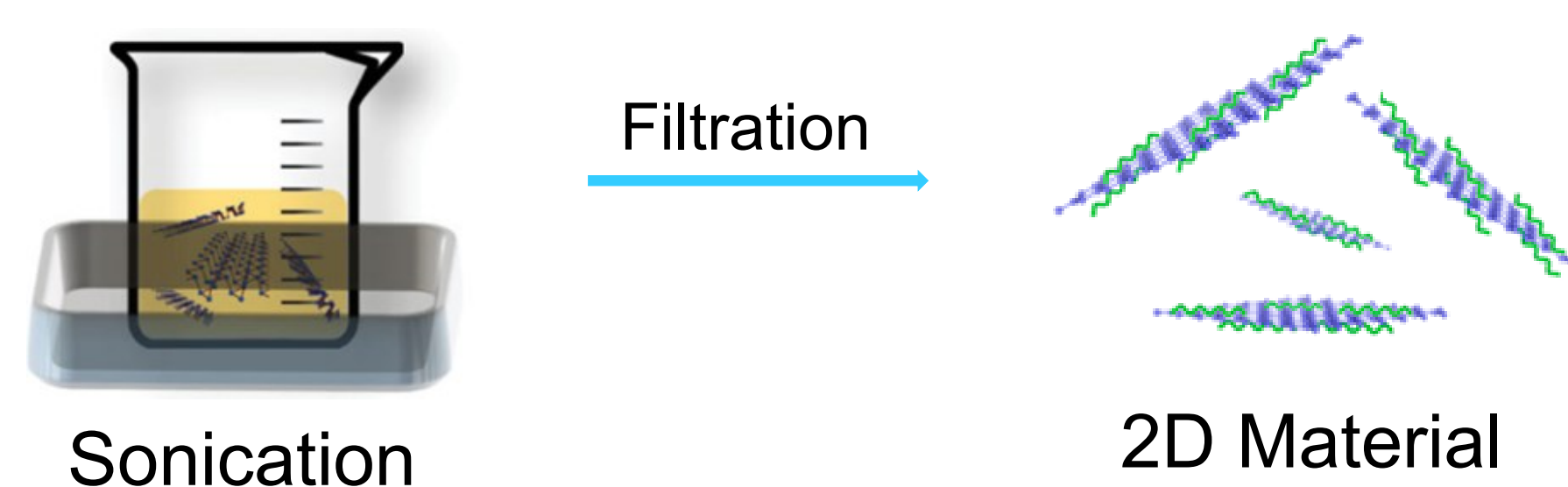
- ❖ Developing two-dimensional (2D) van der Waals (vdW) heterostructures by combining different 2D materials offers new possibilities for battery electrodes by integrating desirable properties and overcoming the limitations of the constituent monolayers.
- ❖ This work is a multidisciplinary collaboration between ATOMS Lab in MIE and LSM Lab in MSE/ChemEng.
- ❖ We are comprehensively investigating C<sub>3</sub>N/black phosphorene (C<sub>3</sub>N/BlackP) heterostructures as anode materials for metal-ion batteries based on DFT calculations and cutting-edge synthesis and characterization approaches.

## Computational details

- ❖ Density Functional Theory (DFT) calculations were implemented with the Quantum ESPRESSO and Quantum ATK packages, utilizing GGA with the PBE parameterization technique.
- ❖ For our investigation, we considered a C<sub>3</sub>N/Black P heterostructure supercell comprised of 3 × 3 × 1 supercells of C<sub>3</sub>N and B.P, with a total of 24 C, 8 N, and 24 P atoms.

## Experimental details

- ❖ The objective is to synthesize few layer heterostructure with minimal defects to enable high performance in aluminum-ion batteries.
- ❖ Liquid phase exfoliation was chosen as the preferred synthesis method due to its scalability and simplicity.



- ❖ The resulting material was characterized using Scanning Electron Microscopy (SEM) imaging; the data obtained will be used for tuning the synthesis process.

## Lattice structure and adsorption of Al ions on C<sub>3</sub>N/BP heterostructure

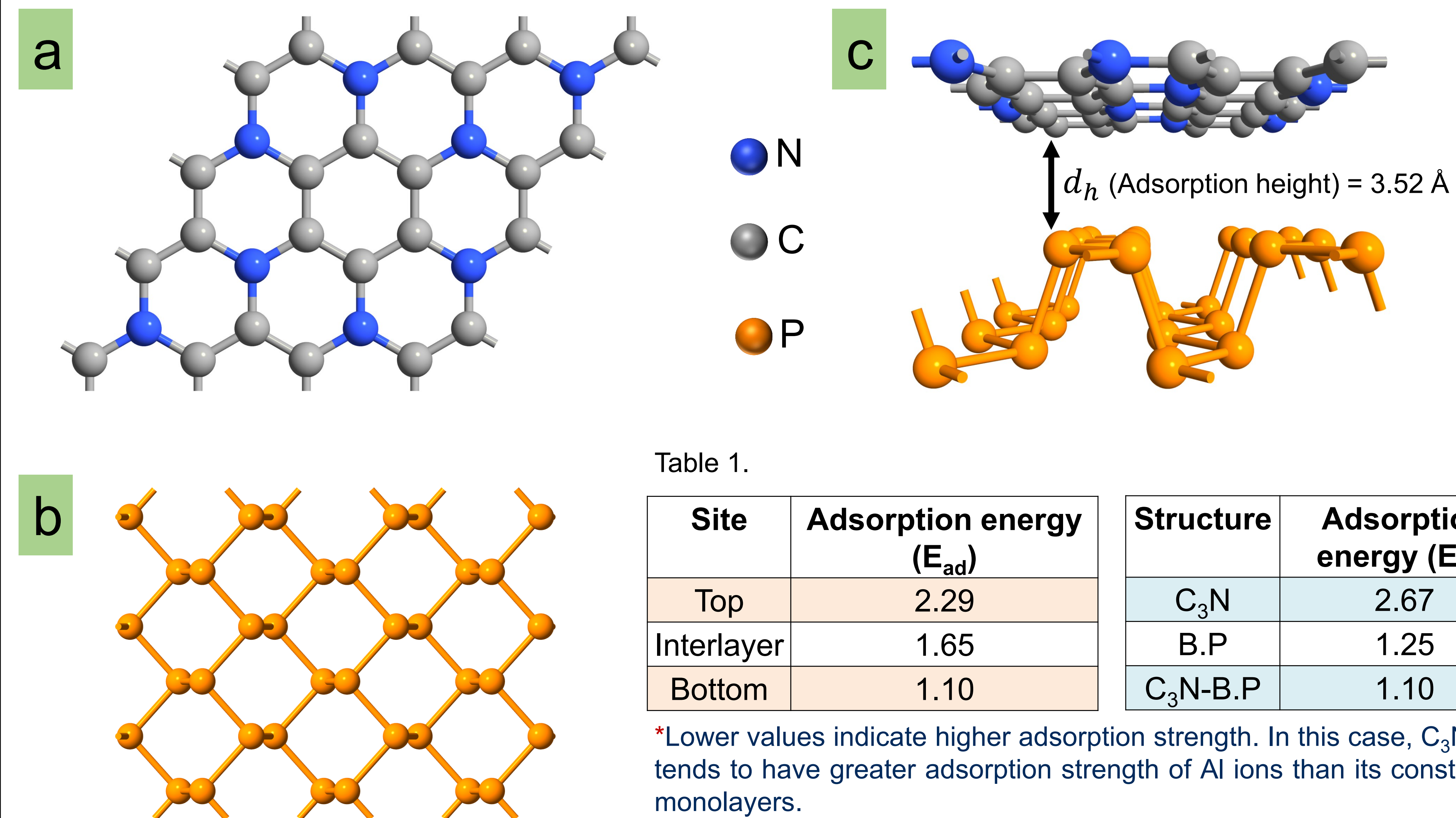


Fig. 1. Diagrammatic representation of monolayers of (a) C<sub>3</sub>N, (b) black phosphorene (B.P) and 2D heterostructure of (c) C<sub>3</sub>N/B.P

Our initial findings indicate that C<sub>3</sub>N/BlackP heterostructures exhibit improved metal-ion adsorption compared to their monolayer counterparts.

## Synthesis of C<sub>3</sub>N/BP heterostructure

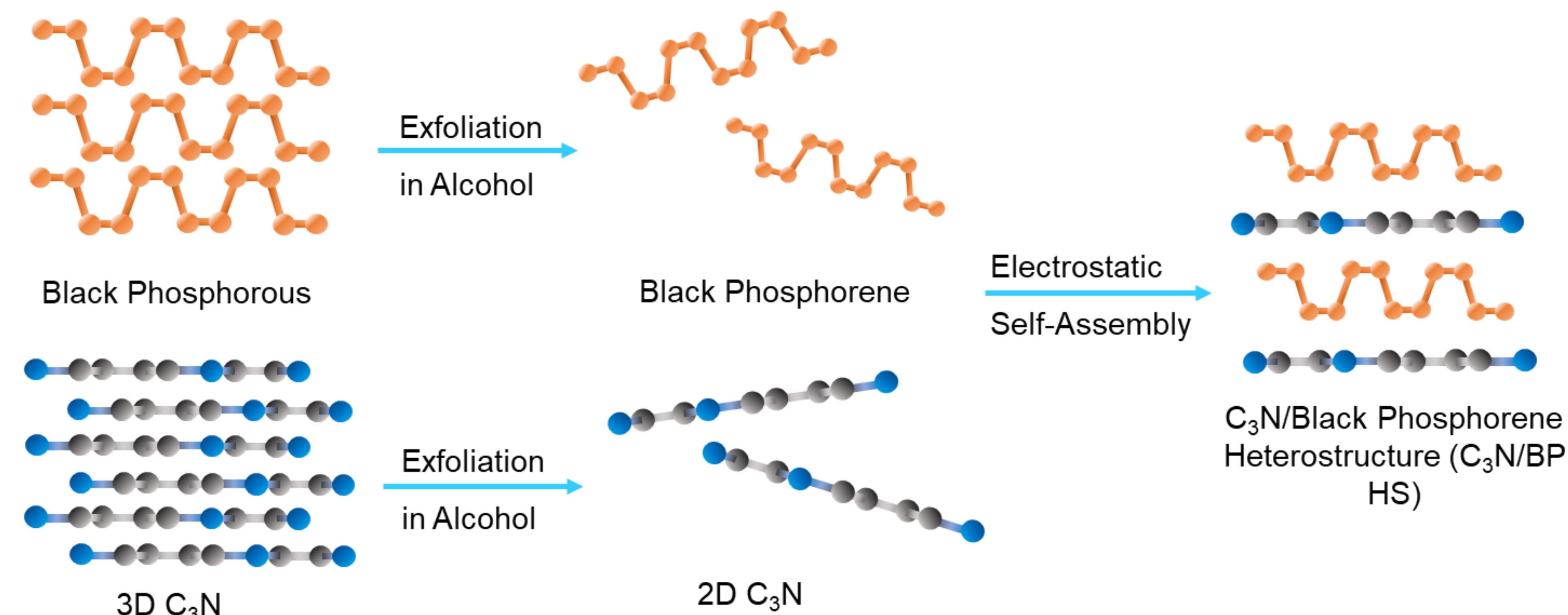


Fig. 2. Synthesis of heterostructure using a 2-step liquid phase exfoliation process

- ❖ This synthesis process can be tuned using the solvent, sonication power and duration of sonication as the parameters.
- ❖ Ideally, the chosen solvent must not react with C<sub>3</sub>N and Black Phosphorous, and the combination of sonication power and duration must deliver enough energy to exfoliate layers of the parent materials.
- ❖ A prototype of the heterostructure was synthesized using a sonication duration of 1 hour at a power output of ~60 W.

## Preliminary Results

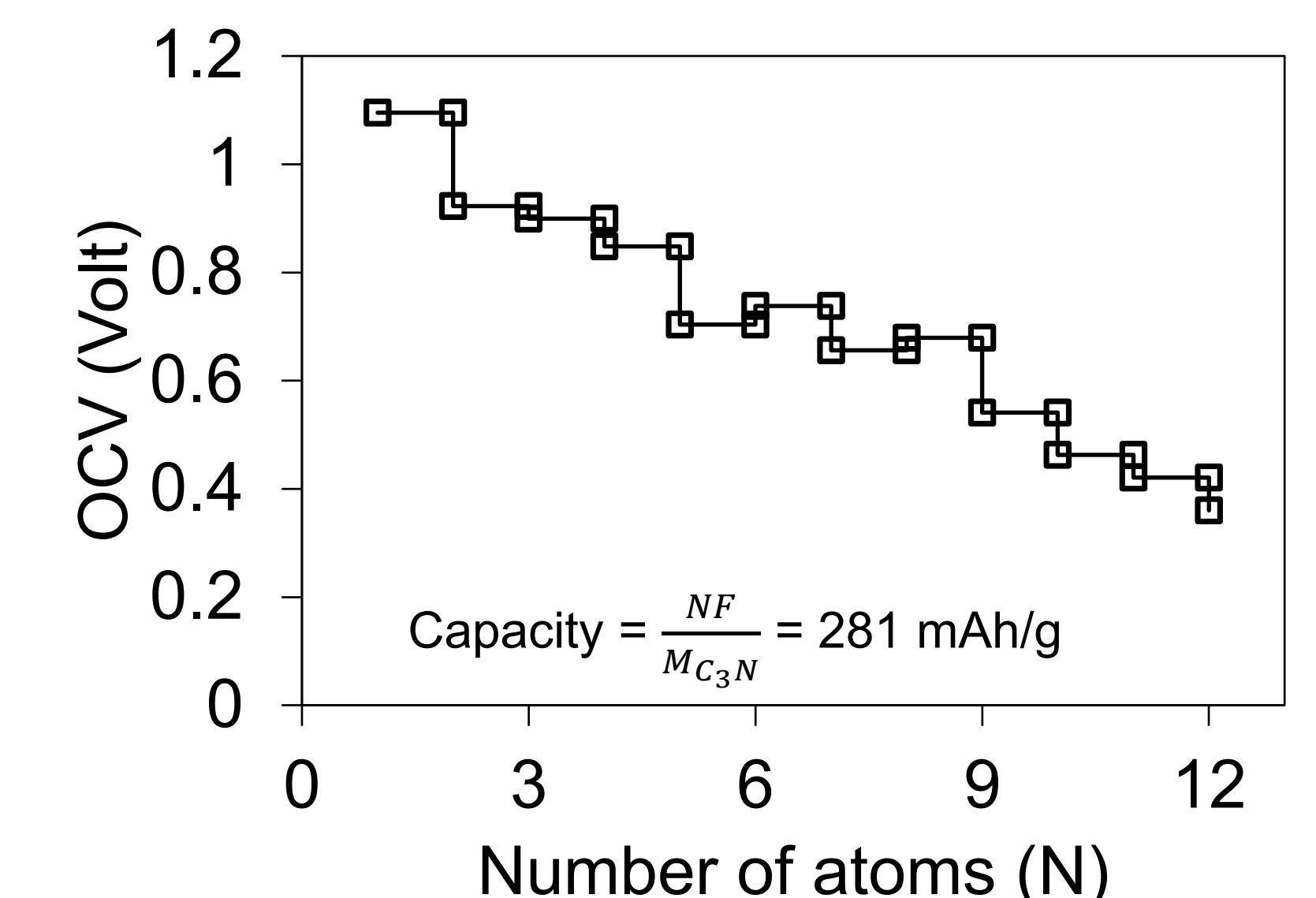


Fig 4. OCV and capacity calculated at 12 Al<sup>3+</sup> ion adsorption

- ❖ DFT predicts an OCV of 0.4 V with a capacity of 281 mAh g<sup>-1</sup>.

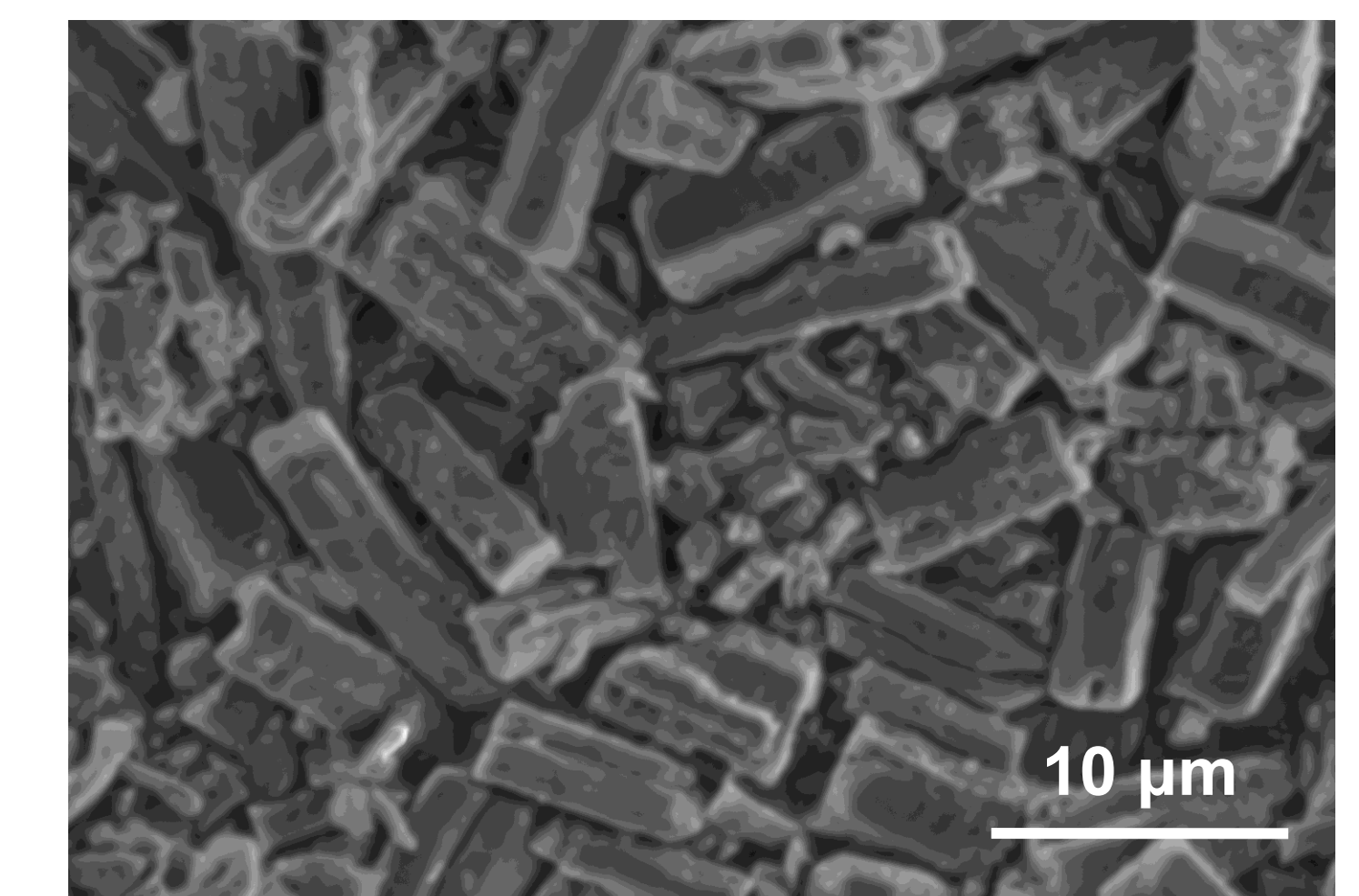


Fig 3. SEM image of synthesised prototype

- ❖ Imaging shows the size of particles is less than 10 μm.
- ❖ Exfoliation must be performed at a higher power output to achieve smaller particle dimensions.

## Conclusion

- ❖ DFT simulations suggest a high-performing heterostructure cathode for Al-ion batteries. However, practical limitations need to be overcome to achieve this performance.
- ❖ The C<sub>3</sub>N/Phosphorene heterostructure prototype was synthesized using liquid phase exfoliation and characterized using SEM imaging.
- ❖ Future work will optimize the synthesis process and use DFT to predict other electronic properties such as the density of states.

## Acknowledgements

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